

# Application of PBBM in risk assessment of effect of acid reducing agents (ARA) on PK and formulation development

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#### **Overview**

- Background to pH-dependent DDI
- Predicting the Effect of Acid Reducing Agents with PBBM
  - *in vitro* methods and PBBM examples
- Roche Case Studies
  - Alectinib
  - Erlotinib
- Future Directions

#### Background



- PK DDIs can cause toxicity or poor efficacy
- Adverse drug events contribute to patient harm and healthcare costs<sup>\*</sup>
- PBPK used to predict and manage metabolic DDIs and inform labels
- Absorption-related DDIs may equal the magnitude of metabolic DDI effects but PBPK is having less impact in this area

\* up to \$177.4 billion annually. Ernst and Grizzle. J Am Pharm Assoc (Wash), 2001. 41(2): p. 192-9

# pH-dependent DDI for Poorly Soluble Weak Bases





#### ketoconazole exposures↓ 95% 2 hours after ranitidine

Piscitelli, S.C., et al. (1991). *Effects of ranitidine and sucralfate on ketoconazole bioavailability*. Antimicrobial agents and chemotherapy. **35**(9): p. 1765-1771.

#### atazanavir exposures↓ 95% after lansoprazole

Tomilo, D. L., et al. (2006). *Inhibition of Atazanavir Oral Absorption by Lansoprazole Gastric Acid Suppression in Healthy Volunteers*. Pharmacotherapy: The Journal of Human Pharmacology and Drug Therapy. 26(3): 341-346.



# **The Effect of Acid Reducing Agents**

 pH-dependent DDI may occur in the stomach when a poorly soluble weakly basic drug with pH dependent solubility is coadministered with an acid reducing agent (ARA) e.g. proton pump inhibitor (PPI), histamine 2 receptor antagonist (H2RA) or antacid



#### **Regulatory View**







Framework for Assessing pH-Dependent Drug-Drug Interactions; Establishment of a Public Docket; Request for Comments

A Notice by the Food and Drug Administration on 05/22/2018

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Notice

# Support for the Value of PBBM



 Multiple published PBBM examples

- Mechanistic studies on the effect of different ARAs
- Proposed biorelevant media

Effect of Gastric pH on the Pharmacokinetics of a BCS Class II Compound in Dogs: Utilization of an Artificial Stomach and Duodenum Dissolution Model and GastroPlus,<sup>TM</sup> Simulations to Predict Absorption

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Utilizing Physiologically Based Pharmacokinetic Modeling to Inform Formulation and Clinical Development for a Compound with pH-Dependent Solubility

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Virtual bioequivalence for achlorhydric subjects: The use of PBPK modelling to assess the formulation-dependent effect of achlorhydria

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# **PBBM to Flag Compounds with High Risk**

Drug Discovery–Development Interface

Prediction of ARA/PPI Drug-Drug Interactions at the Drug Discovery and Development Interface

Stephanie Dodd <sup>1, \*</sup>, Sivacharan Kollipara <sup>2</sup>, Manuel Sanchez-Felix <sup>1</sup>, Hyungchul Kim <sup>1</sup>, Qingshuo Meng <sup>4</sup>, Stefania Beato <sup>5</sup>, Tycho Heimbach <sup>3</sup>

- Integration into GastroPlus of in silico properties or early discovery stage measured properties
- Risk correctly identified for 50% (in silico) & 78% (measured) of studies

# **PBBM Aids Development of Improved Formulations**

Using Absorption Simulation and Gastric pH Modulated Dog Model for Formulation Development To Overcome Achlorhydria Effect

Amitava Mitra,<sup>\*,†</sup> Filippos Kesisoglou,<sup>†</sup> Martyn Beauchamp,<sup>‡</sup> Wei Zhu,<sup>†</sup> Fabio Chiti,<sup>§</sup> and Yunhui Wu<sup>†</sup>

- Free base F1 form. showed reduced AUC with ARA
- F1 compared to F1-1(citric acid & povidone) &
- F2 (HCl salt & citric acid)
- PBBM used to compare Cp(t) for each formulation

USP II, pH 3 NaCL medium at 37 °C Intestinal absorption based on gastric pH 6 with PPI

Mol Pharm 2011 8(6): 2216-2223.



#### Mechanistic Clinical and In vitro Studies

Characteristics of the Human Upper Gastrointestinal Contents in the Fasted State Under Hypo- and A-chlorhydric Gastric Conditions Under Conditions of Typical Drug – Drug Interaction	The impact of reduced gastric acid secretion on dissolution of salts of weak bases in the fasted upper gastrointestinal lumen: Data in biorelevant media and in human aspirates
Studies	Chara Litou <sup>a</sup> , Maria Vertzoni <sup>a</sup> , Wei Xu <sup>b</sup> , Filippos Kesisoglou <sup>b</sup> , Christos Reppas <sup>a,*</sup>
Chara Litou <sup>1</sup> • Maria Vertzoni <sup>1</sup> • Constantinos Gournas <sup>2</sup> • Vassilis Vasdekis <sup>3</sup> • Wei Xu <sup>4</sup> • Filippos Kesisoglou <sup>4</sup> • Christos Reppas <sup>1</sup>	<sup>a</sup> Department of Pharmacy, School of Health Sciences, National and Kapodistrian University of Athens, Zografou, Greece <sup>b</sup> Pharmaceutical Sciences and Clinical Supply, Merck & Co., Inc., Kenilworth, NJ, USA

Pharm Res. 2016 33(6): 1399-1412.

#### • Famotidine and pantoprazole



**Fig. 2** pH in the stomach of fasted healthy adults as a function of time, after administration of 240 ml table water into the antrum of the stomach. Key: White boxes, Phase 1; Light pink boxes, Phase 2; Dark blue boxes, Phase 3. Each box was constructed by using 7–8 individual values.

EJPB 2017 115: 94-101.

- Media for stomach proposed
- Dissolution data shown to be comparable to those obtained with gastric aspirates for 2 test compounds

#### **Incorporation of Dissolution Data in PBBM**

Physiologically Based Absorption Modeling of Salts of Weak Bases Based on Data in Hypochlorhydric and Achlorhydric Biorelevant Media

Filippos Kesisoglou,<sup>1,3</sup> Maria Vertzoni,<sup>2</sup> and Christos Reppas<sup>2</sup>

- Biorelevant GI transfer (BioGIT) measurements with new media used to inform PBBM
- Gastric supersaturation measured and accounted for (can be important for salt forms, acidulant formulations, etc...)
- Successful PBBM simulations for these 2 drugs where solubility vs pH profiles had under predicted the exposures with ARA

#### **Biorelevant** In vitro Tests

Impact of Acid-Reducing Agents on Gastrointestinal Physiology and Design of Biorelevant Dissolution Tests to Reflect These Changes

Domagoj Segregur <sup>1</sup>, Talia Flanagan <sup>2, 3</sup>, James Mann <sup>2</sup>, Andrea Moir <sup>2</sup>, Eva M. Karlsson <sup>4</sup>, Matthias Hoch <sup>5</sup>, David Carlile <sup>5</sup>, Sakina Sayah-Jeanne <sup>6</sup>, Jennifer Dressman <sup>1, \*</sup>

#### In vitro stomach dissolution tests designed for H2-receptor antagonists and PPIs



Figure 3. Typical gastric pH profile without (●) and during chronic PPI administration (■). Meal ingestion is indicated by an arrow.

#### Table 4

The Composition of Media Recommended for Simulating Fasted State Gastric Fluid Under PPI/H<sub>2</sub>RA Co-administration

Component/Parameter	Acetate pH 4 Medium	Maleate pH 6 Medium
Pepsin (mg/mL)	0.1	-
Sodium taurocholate (mM)	0.08	0.08
Phosphatidylcholine (mM)	0.02	0.02
Sodium chloride (mM)	-	22.7
Maleic acid (mM)	-	2.31
Sodium acetate (mM)	33.3	-
+ NaOH (1 M) (mL)	-	qs. (~3.5)
+ HCl (1 M) (mL)	qs. (~25)	
pH	4	6
Buffer capacity (mEq/pH/L)	7.5	1
Osmolality (mOsmol/kg)	91	50
Surface tension (mN/m)	64.49	67.21

Minus sign indicates the absence of component.

### **PBBM to Understand Variability**

Assessment of Bioequivalence of Weak Base Formulations Under Various Dosing Conditions Using Physiologically Based Pharmacokinetic Simulations in Virtual Populations. Case Examples: Ketoconazole and Posaconazole

Rodrigo Cristofoletti <sup>1, 2</sup>, Nikunjkumar Patel <sup>3</sup>, Jennifer B. Dressman <sup>2, \*</sup>

- Model considers surface pH<sub>0</sub> & precipitation
- Self buffering and buffering effects by setting bulk pH to calculated pH0
- Verified with diverse clinical studies (food, cocacola, PPI)
- Sensitivity analysis and virtual BE conducted



#### Alectinib

Physiologically Based Absorption Modeling to Explore the Impact of Food and Gastric pH Changes on the Pharmacokinetics of Alectinib

Neil J Parrott,<sup>1,5</sup> Li J Yu,<sup>2</sup> Ryusuke Takano,<sup>3</sup> Mikiko Nakamura,<sup>4</sup> and Peter N. Morcos<sup>2</sup>

Property	Value
logD	1.96 at pH 3.575
рКа	7.05 base
Permeability	2.5 *10-4 cm/s
Kinetic solubility of granules in	FaSSIF (23 µg/mL)
clinical capsules measured at 37°C	$E_{\alpha}SSIE\left(77\alpha/mI\right)$
in 50 mL of biorelevant media	ressir (// µg/mL)
after 4 hours stirring with a paddle	
speed of 50 rpm . Five milligram of	
RO5424802 was applied.	
Clinical Dose	600 mg



The AAPS Journal (2016)

#### **Model Changes for PPIs**

- Fasting gastric pH increased to 4.5
- Postprandial gastric pH increased to 6.5
- Decreased gastric emptying rate

		Fed				Fasted	
Compartment		pН	Transit Time (h)			pН	Transit Time (h)
Stomach	6.5	4.90	1.00 🔶	1.8 hrs	4.5	1.30	0.25
Duodenum		5.40	0.26			6.00	0.26
Jejunum 1		5.40	0.94			6.20	0.94
Jejunum 2		6.00	0.74			6.40	0.74
lleum 1		6.60	0.58			6.60	0.58
lleum 2		6.90	0.42			6.90	0.42
lleum 3		7.40	0.29			7.40	0.29
Caecum		6.40	4.33			6.40	4.33
Asc Colon		6.80	12.99			6.80	12.99

#### Prediction

- Negligible effect of elevated gastric pH predicted
  - Very limited dissolution in the stomach irrespective of pH
  - Solubility of alectinib at normal healthy gastric pH << 2.4 mg/mL (the value needed for complete dissolution of the dose in a glass of water (600 mg dose / 250 mL))</li>
- Solubility decreases with increased gastric pH have little effect
- This prediction was useful to design pivotal study w.r.t. patient exclusion criteria

# **Confirmation - No Clinically Relevant Effect**



#### Proposal: In such cases PBBM could be sufficient to waive a clinical DDI

E.g.Pabinostat FDA Review Model simulations suggested the lack of effect of elevating gastric pH on panobinostat oral absorption and PK" Ribociclib Bioavailability Is Not Affected by Gastric pH Changes or Food Intake: *In Silico* and Clinical Evaluations

Tanay S. Samant<sup>1</sup>, Shyeilla Dhuria<sup>2</sup>, Yasong Lu<sup>1</sup>, Marc Laisney<sup>3</sup>, Shu Yang<sup>1</sup>, Arnaud Grandeury<sup>3</sup>, Martin Mueller-Zsigmondy<sup>3</sup>, Kenichi Umehara<sup>3</sup>, Felix Huth<sup>3</sup>, Michelle Miller<sup>1</sup>, Caroline Germa<sup>1</sup> and Mohamed Elmeliegy<sup>1</sup>

#### **Erlotinib**

- Lipophilic with high permeability and low solubility
- CYP3A4 & CYP1A2 substrate
- The effect of omeprazole and ranitidine on erlotinib has been studied clinically. Modelling was done retrospectively

Parameter					
logP <sub>(O/W)</sub>	2.7				
pk <sub>a</sub>	5.65				
fu	0.046				
B/P	0.55				
Permeability (cm/s)	caco-2 33.6x10 <sup>-6</sup> -> human Peff 4.3x10 <sup>-4</sup>			4.3x10 <sup>-4</sup>	
	рН		mg/mL		
Duffer colubility of	2.5			0.6	
Burler solubility at	3.4 5 6.5		0.32		
afferent pH (mg/ml)			0.0145		
			0.0058		
Biorelevant solubility	Media	sta	rt pH	end pH	mg/mL
at 37°C (mg/ml)	FaSSIF	(	6.5	6.4	0.0085
	FeSSIF		5	5	0.0533

# **Step 1: Disposition Model**

- Mean Cp(t) for IV and PO crossover study 150-mg tablet vs 25-mg 30 minute intravenous infusion in 20 healthy mainly female subjects
- 2 compartmental model with nonlinear clearance fit gives best fit
- Bioavailability estimated with saturable clearance is 59% vs 106% based on a simple non-compartmental analysis

#### nonlinear model fit in PKPlus



Vc/kg=	0.826	L/kg	CV= 26%
CL2/kg=	0.150	L/h/kg	CV= 54%
V2/kg=	1.138	L/kg	CV= 33%
Vmax =	4.47E-4	mg/s	CV= 53%
Km =	0.232	µg/mL	CV= 77%
K12 =	0.182	1/h	CV= 60%
K21 =	0.132	1/h	CV= 63%
Tlag =	0.228	h	CV= 19%
Ka =	0.731	1/h	CV= 53%
F =	59.27	%	CV= 19%

## **Step 2: Oral Tablet Simulation (Fasted State)**

- Vmax and Km transferred to the enzyme table accounting for changed units and free fraction in plasma
- Default model simulation over estimates observed Cp(t)
- Reduction in %fluid colon improves match



# Step 3: Simulation with/without ARA



#### Without ranitidine





#### With ranitidine



- Stomach pH changed from 1.3 to 4.0
- Gastric transit increased from 0.25h to 0.5h

AUCinf	omeprazole	ranitidine
Observed	54%	67%
Simulated	51%	51%
Cmax	omeprazole	ranitidine
Cmax Observed	omeprazole 39%	ranitidine 46%



#### **Erlotinib Conclusions**

- PBBM indicates high risk of pH-dependent DDI
- Precise prediction of extent of effect is challenging
  - Complex non-linear PK
  - HCl salt dosed
  - Uncertainty in colonic absorption
  - High sensitivity to gastric pH in range 3 5
- Model verification with clinical data recommended
- Subsequent application of a verified PBBM for waiver can be envisaged
  - Formulation changes
  - Different patient populations

# PBBM for ARA DDI Risk & Formulation Development

- PBBM should play a role in integrating physicochemical, in vitro, in vivo and physiological data into a mechanistic framework to yield fuller understanding of pH dependent DDIs
- A bottom-up approach and PSA is useful for early internal decisions
- Multiple PBBM examples support the value during clinical development
- Wider application to streamline drug development and waive unnecessary studies is warranted



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