

The biopharma business of Merck KGaA, Darmstadt, Germany operates as EMD Serono in the U.S. and Canada.

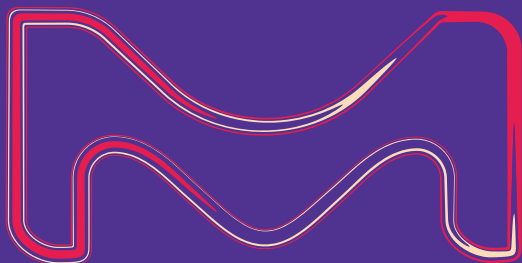
## **Best practices in model development: input of solubility, supersaturation, precipitation and permeability**

Current state and future expectations of Translational Modeling strategies to support drug product development, manufacturing changes and controls

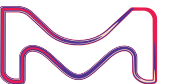
**Christian Wagner**

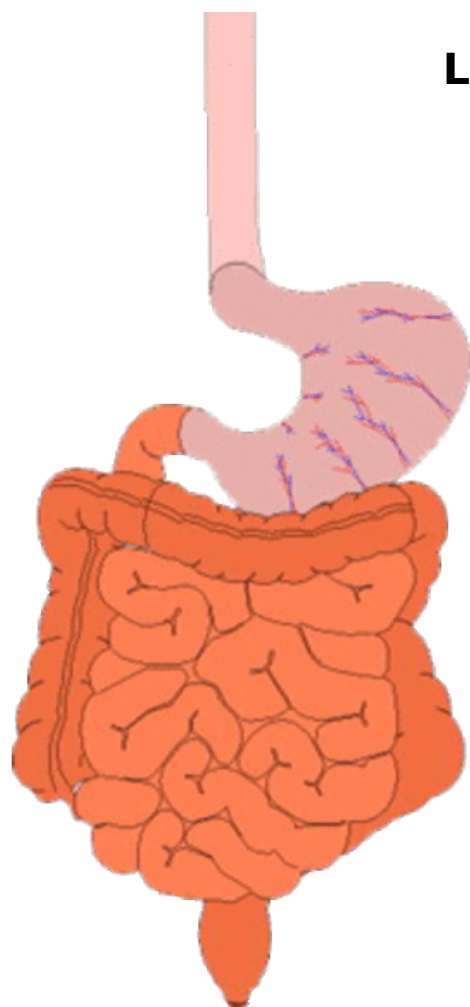
CoE Biopharmaceutics | PharmTech | Chemical and Pharmaceutical Development | Merck KGaA

September 23 - 25, 2019  
College Park, MD, USA



**EMD  
SERONO**





**Liberation and dissolution**



**Absorption**



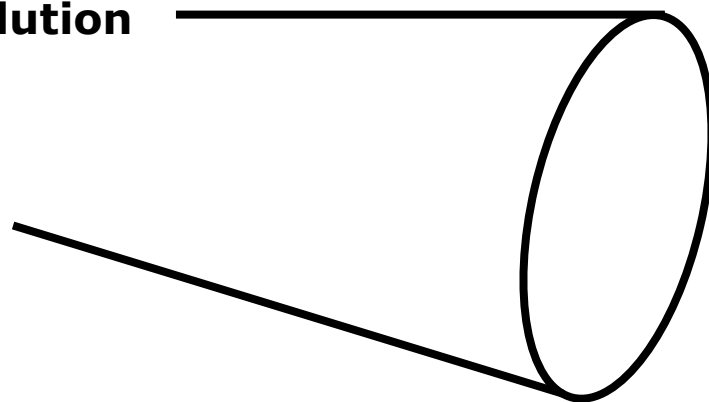
**Distribution**



**Metabolism**



**Excretion**



**Solubility**

**Supersaturation**

**Precipitation**

**Permeability**



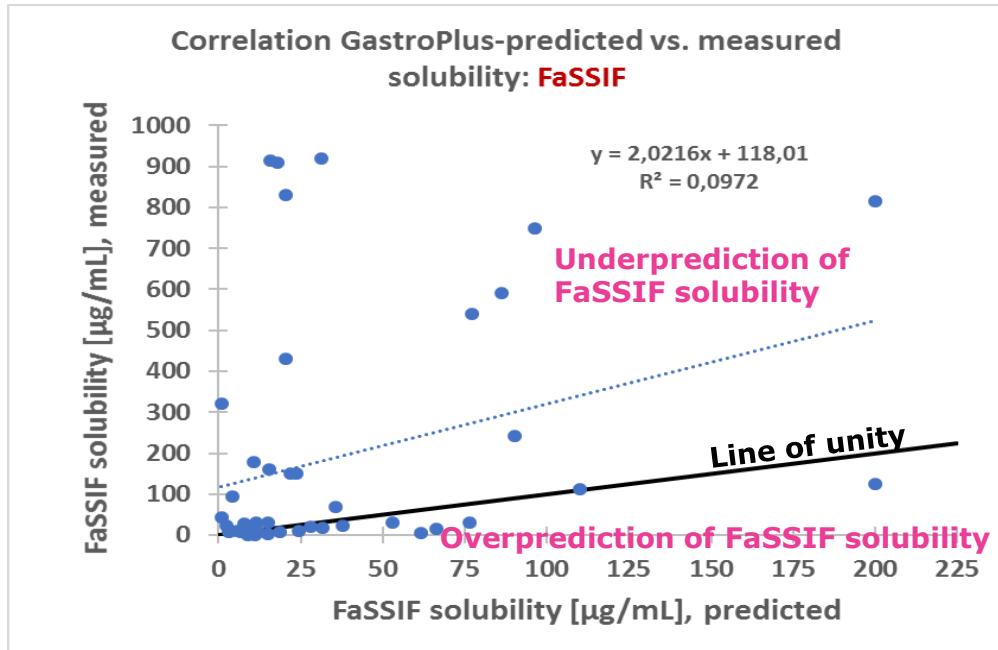


01

## Solubility

# Best practice: Solubility

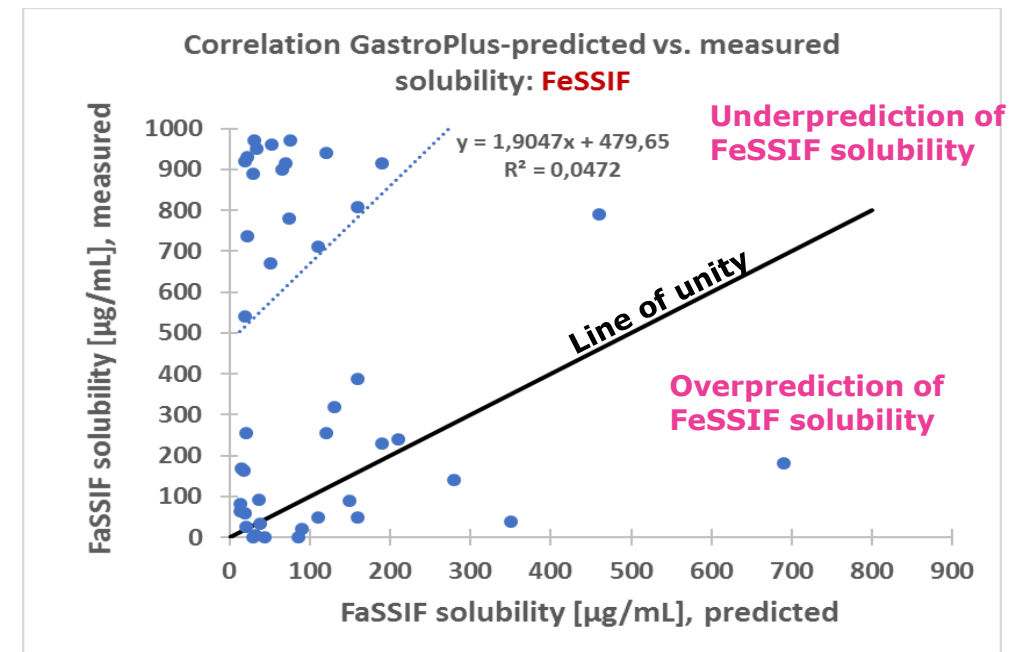
## Case example 1: Predicted vs. observed solubility



### Software-predicted\* vs. measured FaSSIF/FeSSIF solubility

- 46 EMD compounds (LO thru Ph3)
- Mainly weak bases

\* GastroPlus 9.5



## Conclusion

Software tends to underpredict actual FaSSIF/FeSSIF solubility

Use measured solubility as early as possible

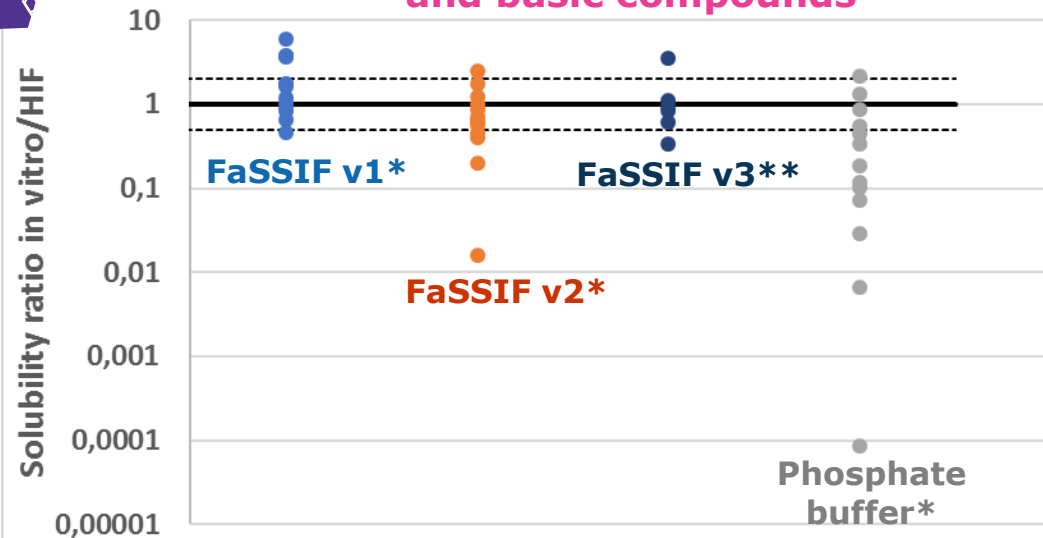
	FaSSIF	FeSSIF
Within 2-fold	33%	9%
Within 5-fold	24%	26%
Outside 5-fold	43%	65%

→ Poor correlation between predicted and measured biorelevant solubility

Best practice: Solubility

## Case example 2: Predictive performance of FaSSIF versions

Solubility ratio (*in vitro*/HIF) of neutral and basic compounds



\* Söderlind et al (2010), MolPharm

\*\* Fuchs et al (2015), EJPB

### Number of cases

	FaSSIF v1	FaSSIF v2	FaSSIF v3	Phosphate
N <sub>total</sub>	13	13	6	13
N <sub>basic</sub>	5	5	2	5
N <sub>neutral</sub>	8	8	4	8

### *In vitro* within 2-fold of observed

	FaSSIF v1	FaSSIF v2	FaSSIF v3	Phosphate
N <sub>total</sub>	8/13 (62%)	7/13 (54%)	4/6 (66%)	3/13 (23%)
N <sub>basic</sub>	3/5 (60%)	1/5 (20%)	1/2 (50%)!	0/5 (0%)
N <sub>neutral</sub>	5/8 (63%)	6/8 (75%)	3/4 (75%)!	3/8 (38%)

FaSSIF v1: Slight trend for over-prediction

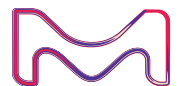
FaSSIF v2: Slight trend for under-prediction

FaSSIF v3: Seems ok (but many outliers of FaSSIF v1 and v2 not included in FaSSIF v3 analysis)

Phosphate: Pronounced under-prediction

## Conclusion

Based on literature data, FaSSIF v1 seems to be most predictive for *in vivo* solubility in HIF. For FaSSIF v3, the basis of data is even smaller.



# Best practice: Solubility

## Case example 3: "Literature says"\*

### REVIEW

#### Predictive Performance of Physiologically Based Pharmacokinetic Models for the Effect of Food on Oral Drug Absorption: Current Status

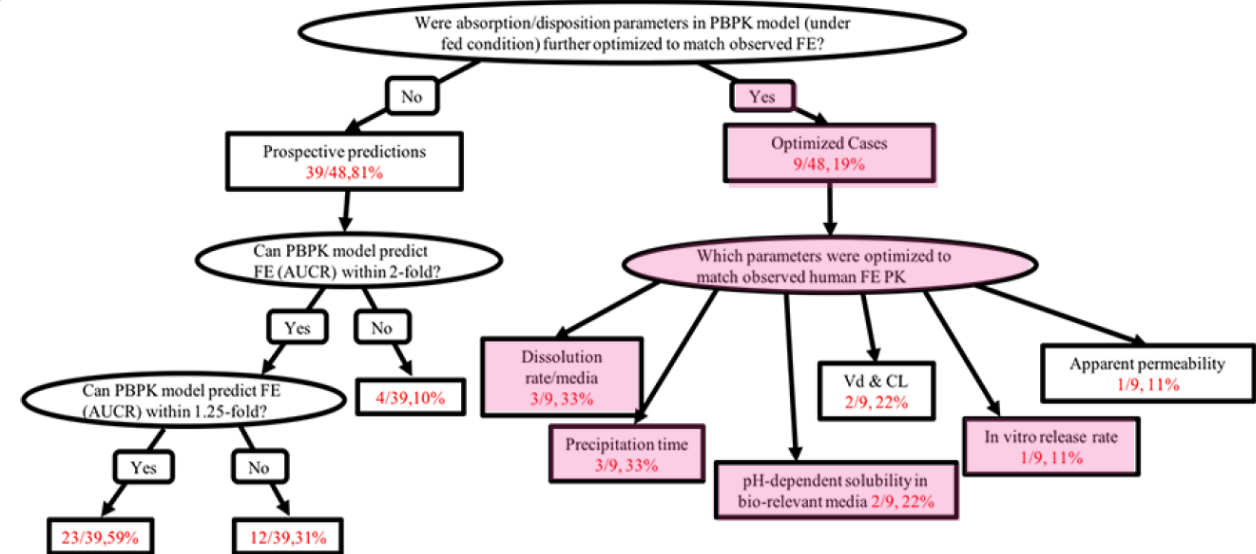
Mengyao Li<sup>1,2\*</sup>, Ping Zhao<sup>1,3</sup>, Yuzhuo Pan<sup>4</sup> and Christian Wagner<sup>1,5</sup>

CPT Pharmacometrics Syst. Pharmacol. (2017) 00, 00; doi:10.1002/psp4.12260; published online on 0 Month 2017.

- 15 peer-reviewed publications and 2 FDA reviews
- 27 compounds (22 basic; 2 acidic; 2 ampholytes; 1 neutral) with 48 food effect simulations
- 63% poorly soluble, 15% highly soluble, 22% n.a.
- $F_a < 50\%$  in approximately 50% of cases
- 81%: Prospective simulations; 19% optimized fed parameters

## Conclusion

**Evidence for dis-connect between *in vitro* and *in vivo* solubility. Physiology in PBPK tools needs improvement.**



- Fitted parameters if fed simulation did not fit to observed data: Dissolution/release rate, precipitation parameters, solubility.
- "Input parameters for both absorption and disposition models could be obtained from various sources, reflecting a lack of standardization and large variability in the quality of input data."
- Physiology not always reflected well in PBPK software tools, leading to mis-prediction of site-specific solubility (fed gastric physiology; dynamic decrease of pH over time in fed stomach; liquid volumes in colon)

\* Li et al (2017), CPT-PSP

# Best practice: Solubility

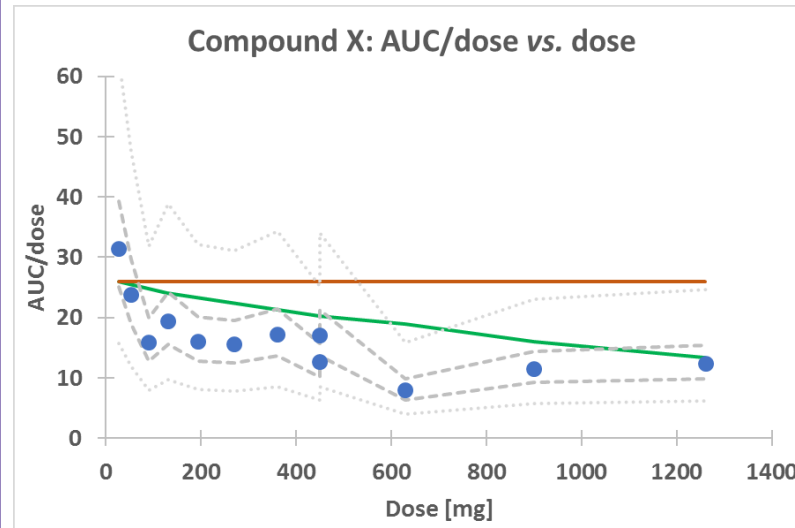
## Case example 4: Ion-effect

EMD compound X shows **ion-effect**, with significantly **impaired solubility** in the presence of **physiological chloride concentration**.

pH	Solubility of compound A [mg/mL]	
	Without Cl <sup>-</sup>	With 100 mM Cl <sup>-</sup>
1.1	~ 1	~ 0.01
4.5	~ 0.7	~ 0.007
6.8	~ 0.04	~ 0.004

Use PBBM to 1) **set specs for PSD** and 2) **de-risk BE study**. But: **Which solubility input to use?**

### What is the impact on simulations...



### ... and decision-making?

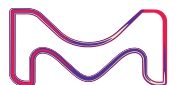
#### BCS class 4

- **Ion effect considered:** BE correctly predicted, formulation development de-risking successful
- **Ion effect not considered:** Overprediction of absorption, successful BE prediction jeopardized

## Conclusion

**Test if compound shows ion effect. If so: Take effect into account, especially if the compound is poorly soluble.**

Experience for comparably **highly soluble compounds** (BCS 1; DCS 1/2a): **No pronounced impact** if ion effect is reflected in solubility input (D/S ratio reasonably low even in presence of counter ions).



# Best practice: Solubility

## Case example 5: Impact of puffer species on solubility of pazopanib

IQ Consortium

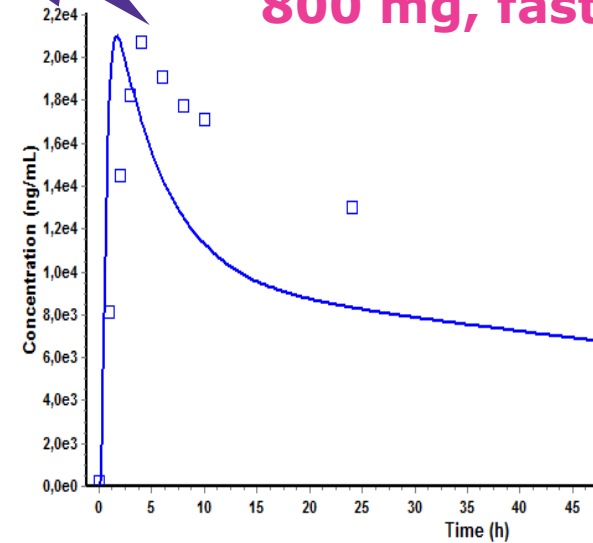
### Pazopanib

pKa	~ 2 (b); ~ 6 (b)
Solubility [µg/mL]	~ 1 (FaSSIF), ~ 3 (FeSSIF)
Permeability	High (Caco2)
Question	FIH: Absorption limitations?
BA (fasted)	~ 25%
Absorption	Dose-dependent; impaired (fasted)
Food effect	~ 2-fold @ 800 mg
Question	Predict food effect

### Modeling strategy

- Absorption model: Measured solubility; Caco2 permeability. Fit precipitation time to match PK from SAD.
- Post-absorptive DD: From IV data

### 800 mg, fasted



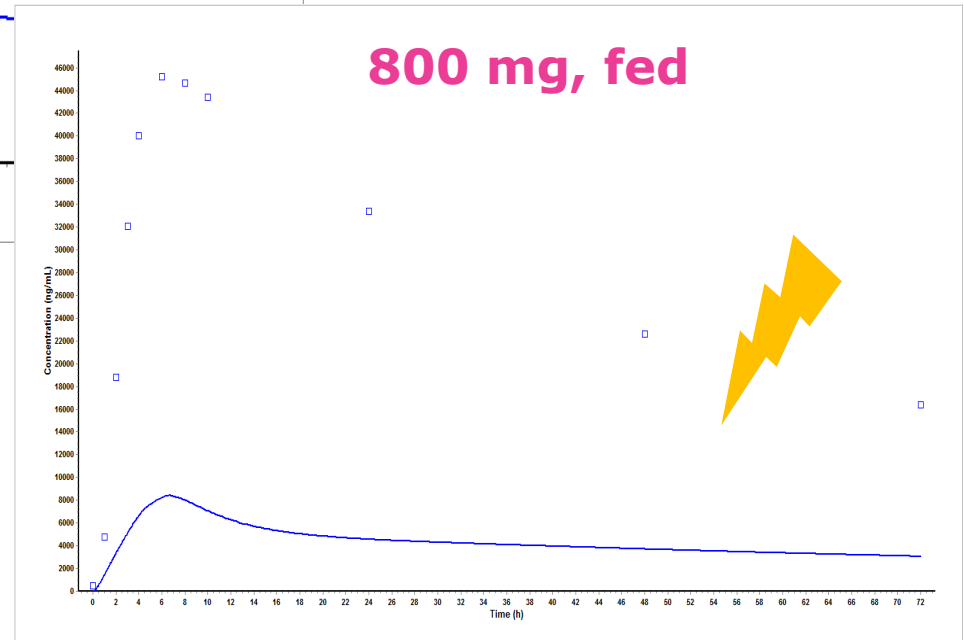
### Fasted

- AUC, Cmax, tmax within 2-fold of observed data
- Shape ☹️
- Successful verification?

### Fed

- Negative food effect predicted (low solubility; shortcomings software)
- Predictivity of FaSSIF/FeSSIF solubility for pazopanib?

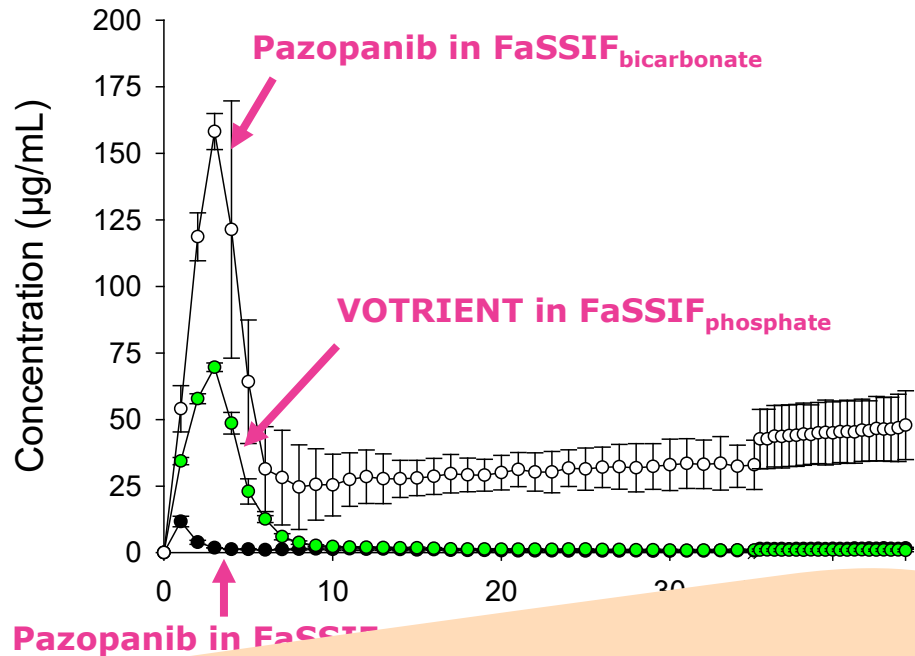
### 800 mg, fed



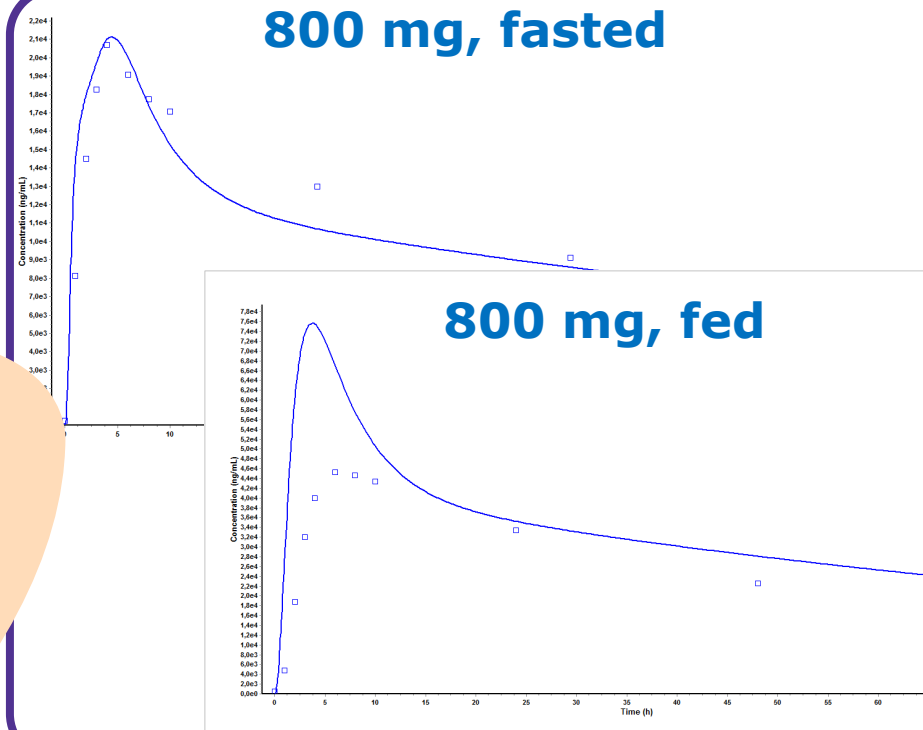
# Best practice: Solubility

## Case example 5: Impact of puffer species on solubility of pazopanib

Transfer model and the use of a more biorelevant bicarbonate buffer\*,\*\*



- Physiological buffer species in gut: Bicarbonate, not phosphate\*\*
- Very pronounced increase in solubility and supersaturation when going from FaSSIF<sub>phosphate</sub> to FaSSIF<sub>bicarbonate</sub>.



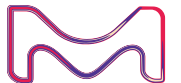
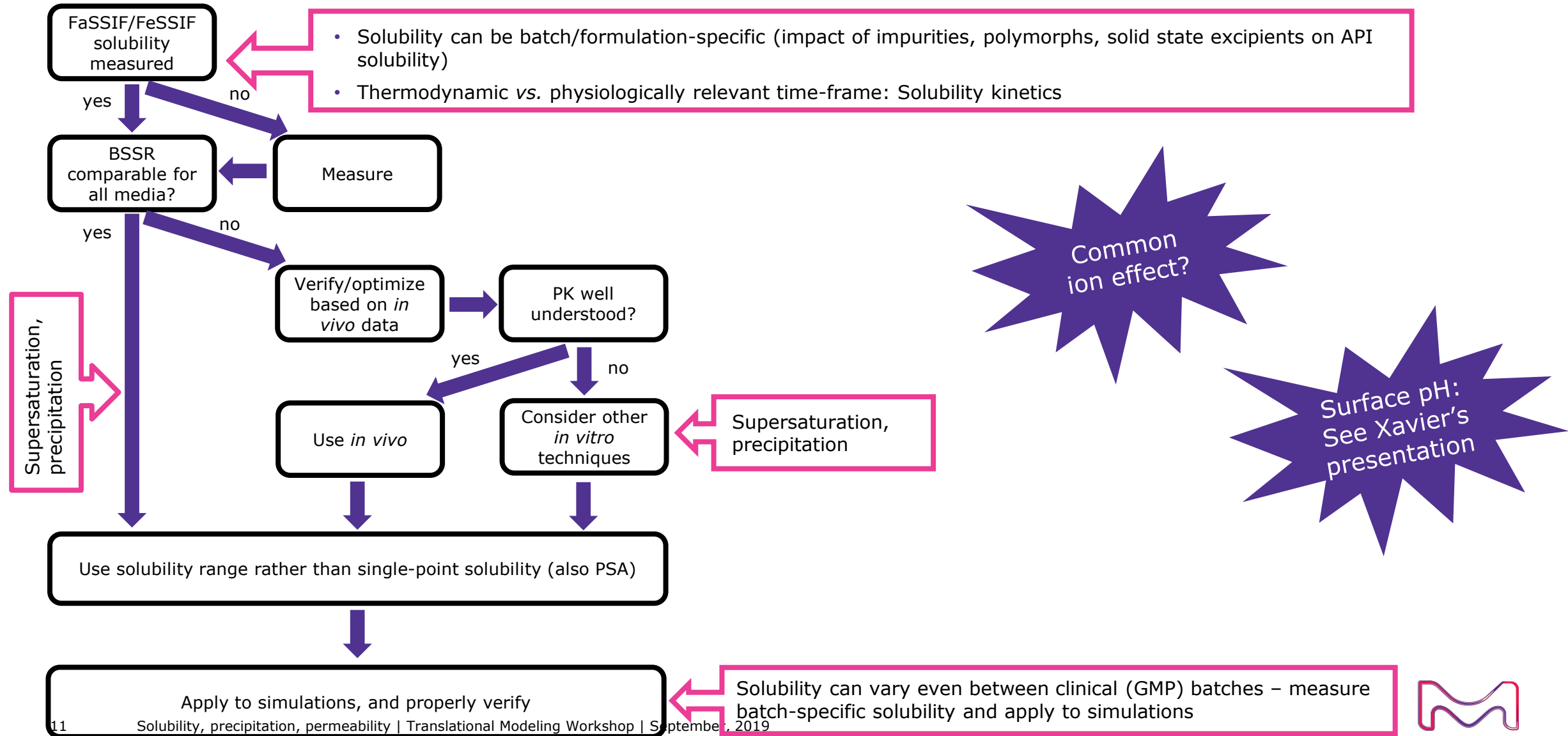
**Increased solubility and supersaturation from transfer experiments in FaSSIF<sub>bicarbonate</sub>**

- **Fasted:** AUC, C<sub>max</sub>, t<sub>max</sub> within 2-fold, acceptable profile shape
- **Fed:** Slight overestimation of FE, but trend and extent of FE were predicted well.

### Conclusion

For challenging compounds, a more biopredictive solubility medium / setup may yield better simulation results

# Best practice: Solubility Summary





02

## **Supersaturation and Precipitation**

# Best practice: Supersaturation and precipitation

## Methods to predict precipitation

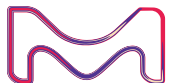
### *In vitro* assays

- Dumping experiments [Kambayashi et al (2016) EJPB; Jakubiak et al (2015) MolPharm]
- Two-compartmental transfer model [Kostewicz et al (2004), JPharmPharmacol] and variations thereof [e.g., Jede et al (2018), JPharmPharmacol]
- Multi-compartmental models, also accounting for drug absorption [e.g., Gu et al (2005), JPharmSci; Psachoulis et al (2012) PharmRes; Kourentas et al (2016) EJPharmSci]
- Bi-phasic precipitation experiments [e.g., Tsume et al (2018) JPharmSci]
- TNO model (TIM-1, TIM-2)

See Ed's presentation for more details

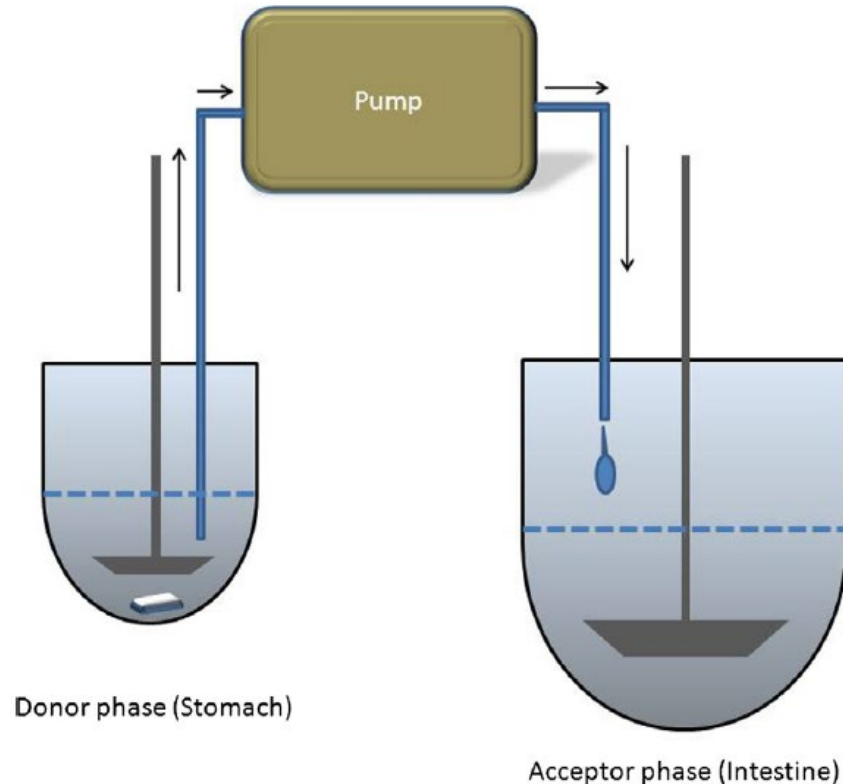
### *In silico* approaches

- Approaches based on classical nucleation and crystal growth theory [e.g., Carlet et al (2010) PharmRes]

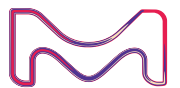
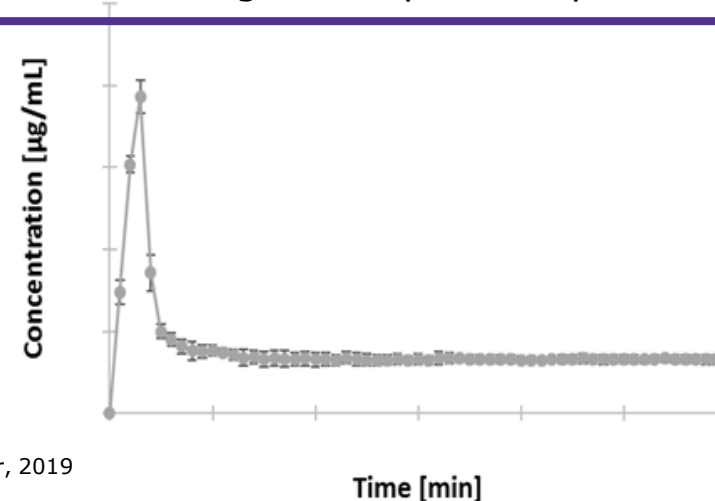


# Best practice: Supersaturation and precipitation

## Reminder: Transfer model



- Example for precipitation setup which **does not take absorption into account**
- Two-compartmental model, based on USP 2 apparatus [Kostewicz et al (2004), JPharPharmacol], or mini-scale [Jede et al (2018), JPharPharmacol] for preformulation application
- Simulated stomach (donor), containing SGF (or versions thereof)
- Simulated intestine (acceptor), containing FaSSIF (or versions thereof)
- Transfer of API solution/suspension from “stomach” to “small intestine” via pump, and measurement of concentration of dissolved drug in acceptor compartment



Best practice: Supersaturation and precipitation

## Case example 1: Transfer model to simulate precipitation of Cpd A

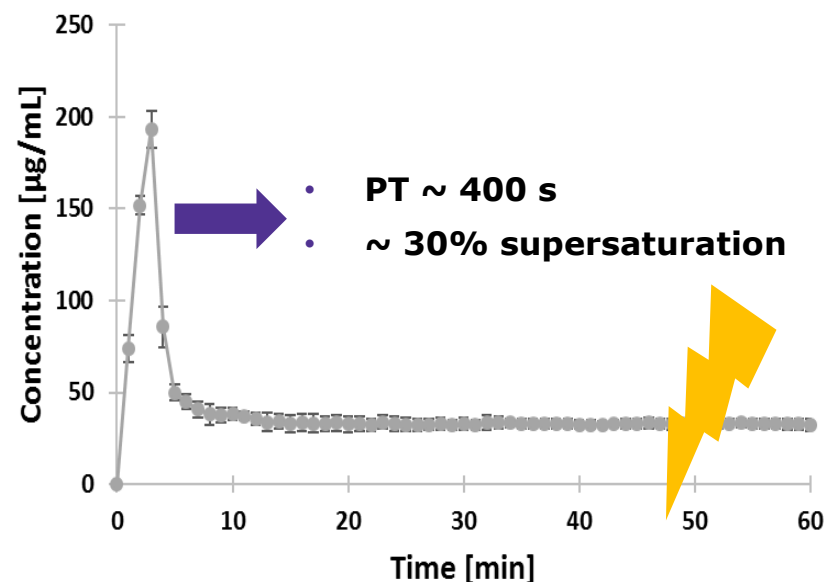
### Compound "A", EMD Serono

pKa	~ 6 (b)
Solubility [ $\mu\text{g/mL}$ ]	~ 30 (FaSSIF)
Permeability	High
Question	FIH: Absorption limitations?

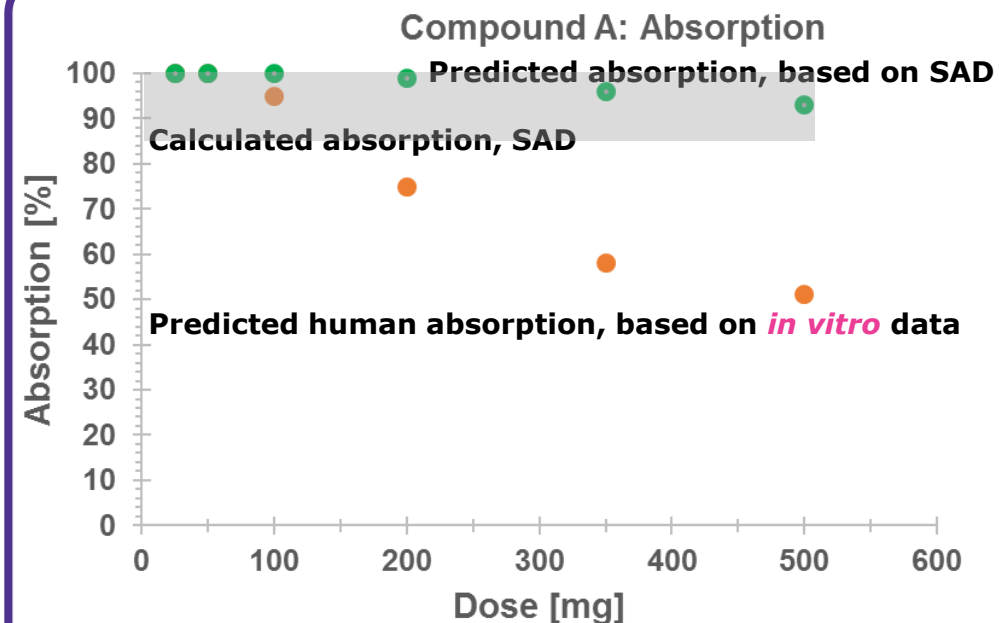
### Scaling of precipitation parameters from *in vitro* data

- **Transfer model**: Pronounced underprediction of absorption at medium/high doses, **assay not predictive** (lack of absorption sink)

### Small-scale transfer model



**PBBM**



Best practice: Supersaturation and precipitation

## Case example 2: Transfer model to simulate precipitation of Cpd B

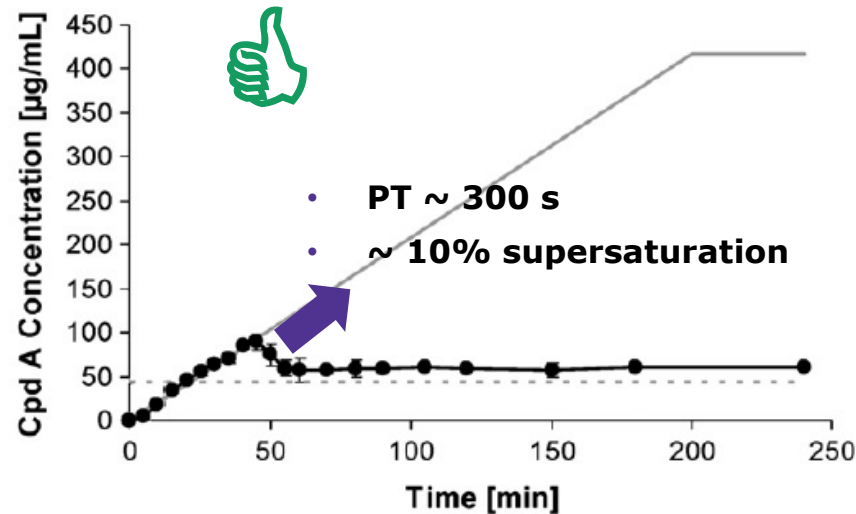
### Compound "B", Merck & Co

pKa	~ 4 (b); ~ 8 (b)
Solubility [ $\mu\text{g/mL}$ ]	~ 50 (FaSSIF)
Permeability	Low (BCS 4)
Question	Precipitation impacts absorption?

Wagner et al (2012), EJPB

### Scaling of precipitation parameters from *in vitro* data

- **Transfer model**: Good prediction of Compound B PK profile using *in vitro* precipitation data as input parameter for the model
- **Lack of absorption sink not deemed critical**, as compound yields low permeability anyway

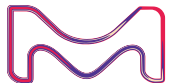


PBBM



## Conclusion

When using data from *in vitro* setups like the transfer model: Permeability matters...



# Best practice: Supersaturation and precipitation

## Case example 3: FIH absorption prediction, optimization based on animal PK

### Compound "C", EMD Serono

pKa	~ 5 (b)
Solubility [µg/mL]	~ 3 (FaSSIF); < LOQ @ pH > 4 → ASD
Permeability	High
Question	Predict FIH absorption

### Initial approach

- Build **bottom-up** model, using **measured** pH-dependent and biorelevant solubility
- Dumping

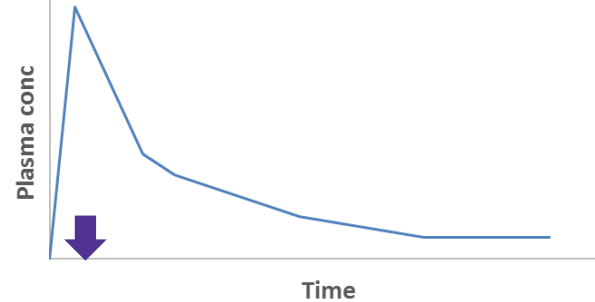
### Conclusion

**Good extrapolation from animal PK to human, also for many other compounds.**  
**Cave: Species differences**

**Complex formulation, high uncertainty**

**Simulate range rather than single value**

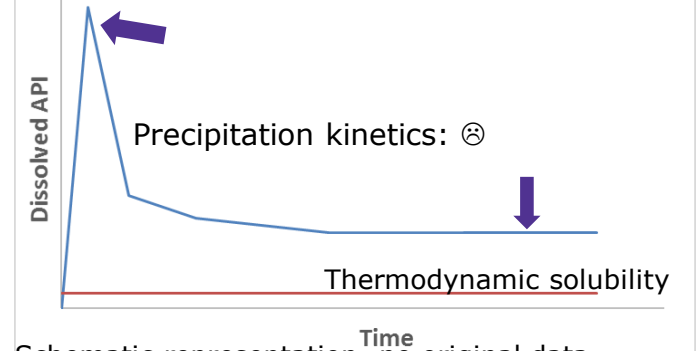
### Animal PK: Dissolution



Schematic representation, no original data

**Early t<sub>max</sub>** across species and dose levels: **Fast dissolution from ASD**, **precipitation** to govern fraction dissolved.

### Transfer model: Supersaturation

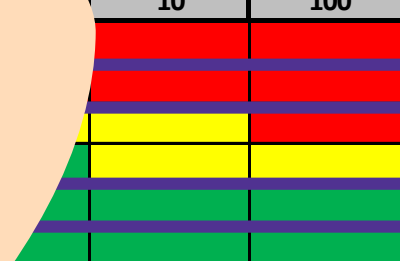


Schematic representation, no original data

**Which value to use?** Peak concentration or stable supersaturation? Conservative estimate used.

### at dose [mg/kg]

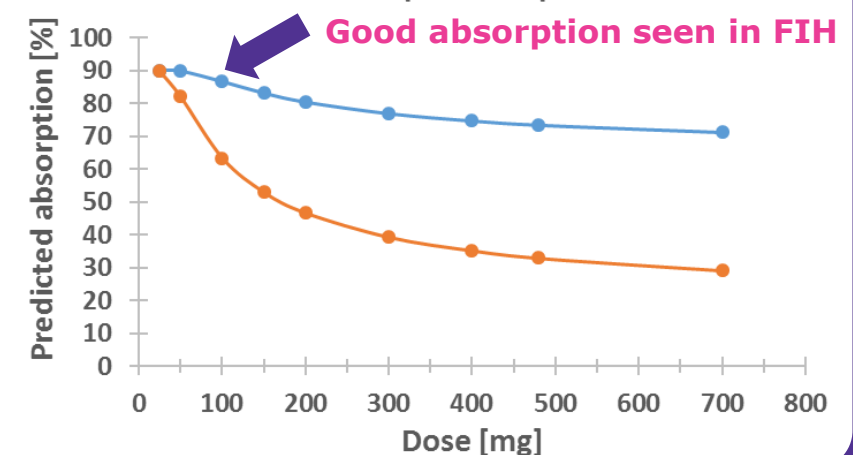
10 100



precipitation time which fits  
dissolved exposure (AUC, C<sub>max</sub>,  
t<sub>max</sub>) for each species and dose level

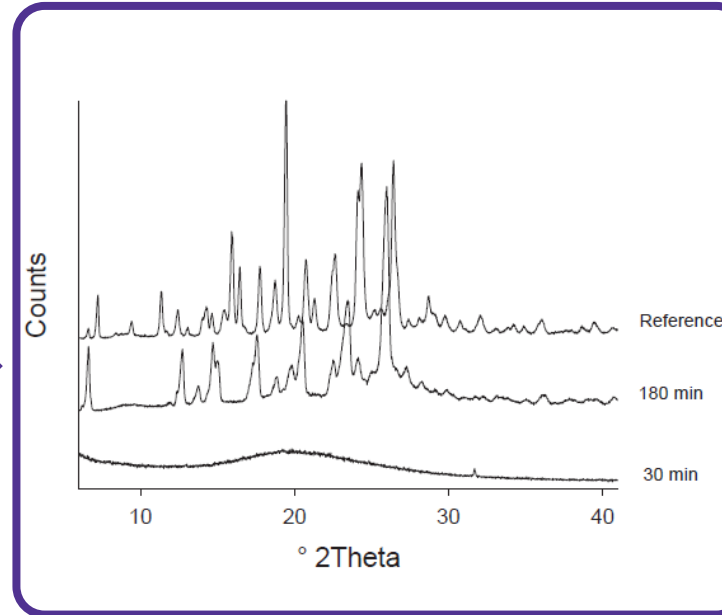
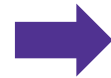
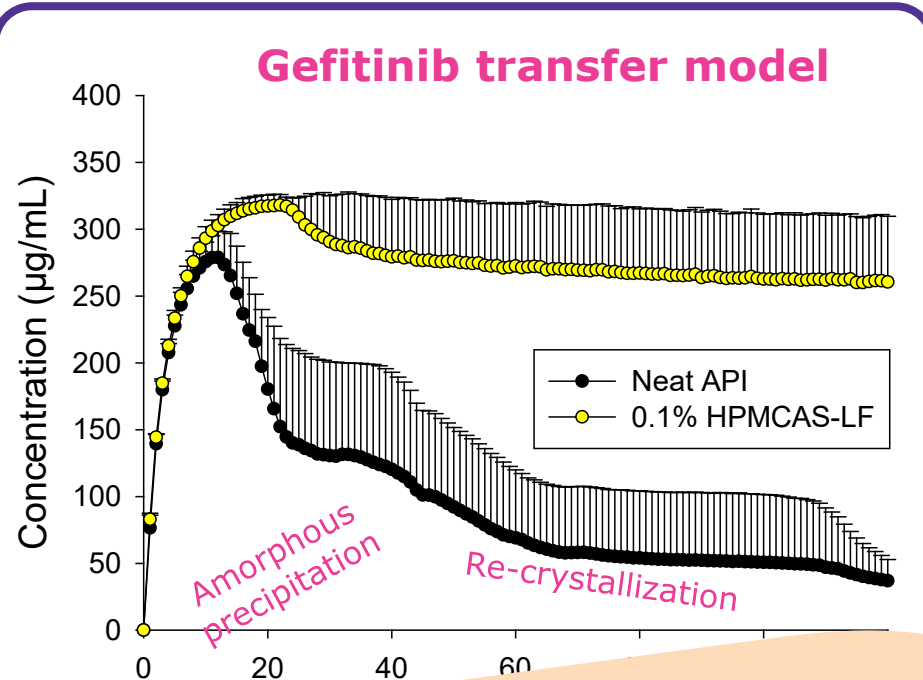
Example: Rats  
10th September 2019

### Predicted absorption of Cpd C in fasted humans



# Best practice: Supersaturation and precipitation

## Case example 4: Precipitate $\neq$ precipitate



- Multi-phasic precipitation
- Amorphous precipitate confirmed after 30 min (intermediate supersaturated state)
- Slow re-crystallization

## Conclusion

**Precipitate  $\neq$  precipitate.**  
**Characterization of precipitates, incl. re-dissolution, is thought to improve predictive performance for challenging compounds.**

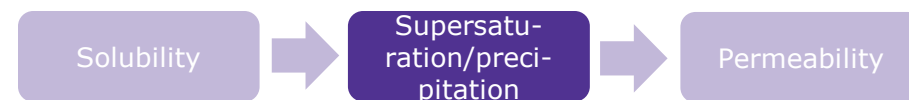
Presence of amorphous precipitates confirmed using transfer model

Gefitinib shows very high PK variability. Reasons not yet clear (genetic CYP polymorphism; gastric emptying; precipitation?)

- Though no PBBM was conducted for gefitinib, relevance of *in vitro* results for absorption modeling is deemed to be high (supersaturation; decreased precipitation)
- Only very few cases where amorphous precipitation of weak bases has been described in the literature. Future investigations needed.

# Best practice: Supersaturation and precipitation

## Summary



- **Some approaches reported in literature, most of them describing good predictivity of the assay.**
- **No systematic evaluation available.**
- **Internal experience with *in vitro* setups to predict precipitation: Mixed. Publication bias?**

**Supersaturation: *In vivo* solubility often higher than what is measured *in vitro*. Supersaturation observed in *in vitro* setups seems to be a more predictive value than (thermodynamic) solubility.**



**Best practice?**

**Precipitation kinetics: Limited confidence in predictive performance of *in vitro* precipitation models**

**Optimizing precipitation parameters based on *in vivo* data currently seems to be the approach with the highest confidence.**





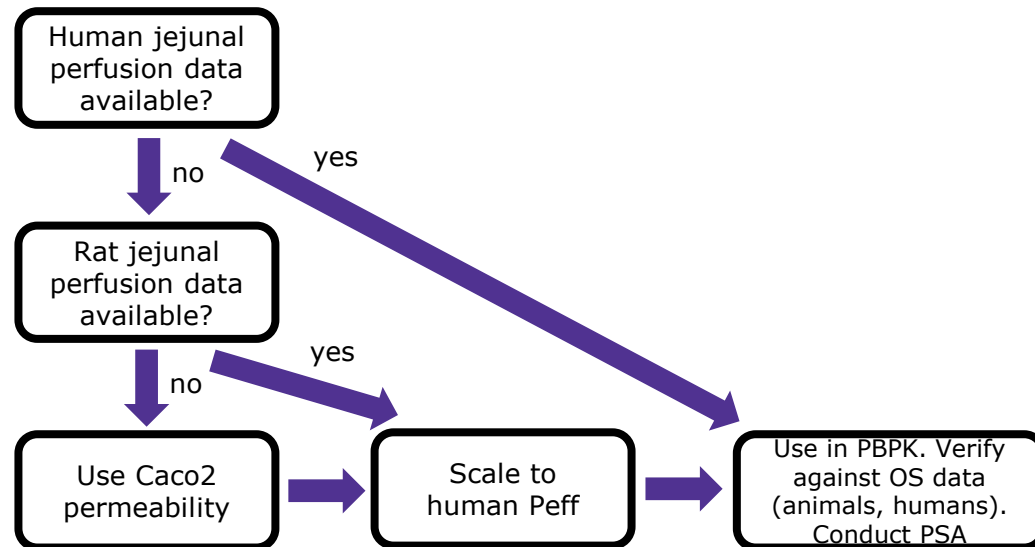
03

## Permeability

# Best practice: Permeability

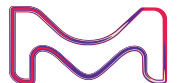
## Passive, transcellular absorption

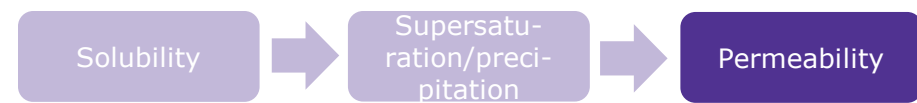
- Presumably most accurate estimate for permeability: (Rat) jejunal perfusion, which is also accepted for BCS classification [e.g. Kesisoglou (2013) AAPSJ]
- Caco2 assay (other cells; PAMPA) standard assay in industry. Important: Scaling from *in vitro* data to *in vivo* situation, using set of calibrants →  $P_{eff}$



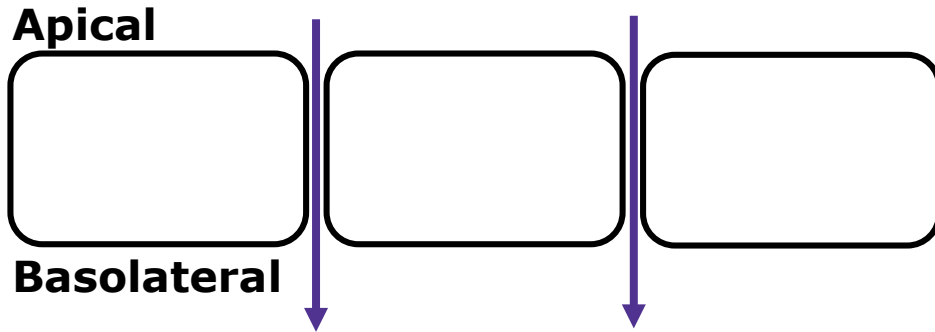
### Open questions, current gaps

Level of "accuracy"	How much do we need when permeability is high anyway?
Optimization of permeability	How to optimize when no OS available / dissolution is assumed to decelerate absorption?
Food/excipients	How to model?
Drift of particles into UWL	Is permeability a function of dose? [Sugano (2013) IntJPharm]
Saturable active transport	Low basis of data. Lack of experience. Scaling from <i>in vitro</i> to <i>in vivo</i> .
Colonic absorption	Overestimated? [Kesisoglou (2013) AAPSJ]





## Best practice: Permeability Paracellular absorption



**Paracellular absorption:** Considered to contribute only marginally to overall absorption

- Often small, hydrophilic molecules
- Very **few reports** describing how paracellular absorption is handled in PBPK, thus limited experience
- Rule of thumb [Peters (2008) CPT]

### Conclusion

- **More experience needed**
- **"Best practice": ?**

#### ☒ Include Paracellular Permeability

Paracellular model:

Molecular Radius (Å):

Human  
Paracellular Jejunal Peff (cm/s x 10<sup>4</sup>):   
Transcellular Jejunal Peff (cm/s):

#### Total Jejunal Permeability

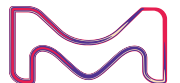
Source:   
Peff (cm/s x 10<sup>4</sup>):   
Sim Peff x 10<sup>4</sup> (Human):

#### Physiological Parameters

Intestinal Electrical Potential:

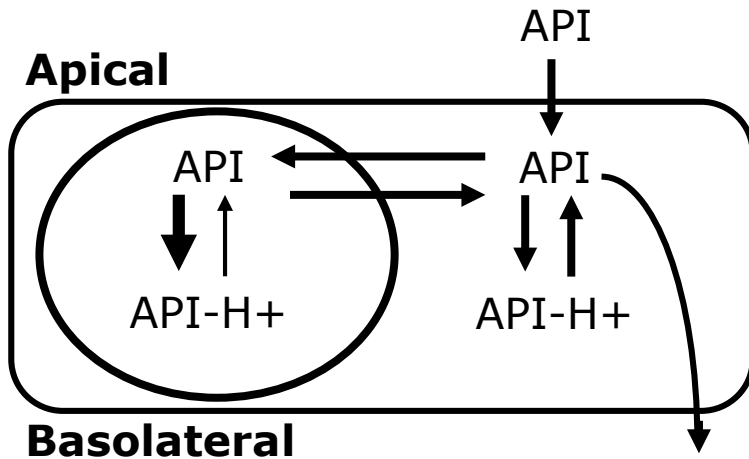
Example: GastroPlus 9.7

- If compound is assumed to have paracellular absorption: Use build-in calculator (input needed: Calculated Peff; molecular radius)
- **Internal experience:** Paracellular absorption usually not taken into account.

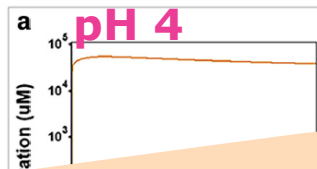


# Best practice: Permeability

## Lysosomal trapping



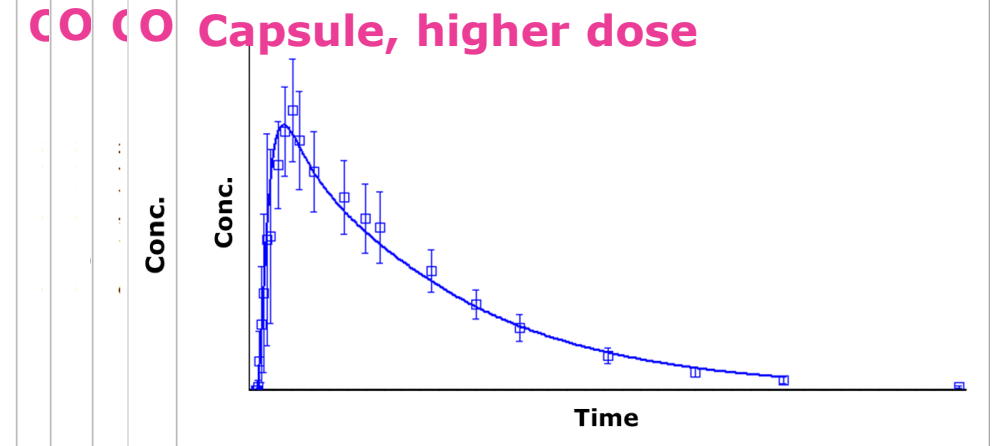
- Trapping of **lipophilic, ionizable** compounds described for several compounds (antipsychotics; TCA; SSRI; dextromethorphan, amodiaquine)



## Conclusion

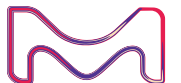
- Differentiation low permeability vs. trapping**
- Semi-mechanistic simulation of lysosomal trapping seems feasible**

- EMD compound C: Lipophilic base
- Late  $t_{max}$  values (6 – 10 h), **independent** from formulation (OS, capsule, tablet)
- Model building to set dissolution specifications for BE study



## Observations:

- Oral solution:  $C_{max}/t_{max}$  **not captured** well
- Optimizing permeability did not yield good results ( $t_{max}$  too early or absorption too low)
- Added **lysosomal trapping** into the model: Decrease  $f_{u,ent}$  from 100% (default) to 50% (fitted) to slow down mass transfer from enterocytes into systemic circulation [Bolger et al (2019), JPharmSci].
- Good fit of data for oral solution. Model was subsequently **used successfully to simulate PK for various oral formulations, incl. BE study.**





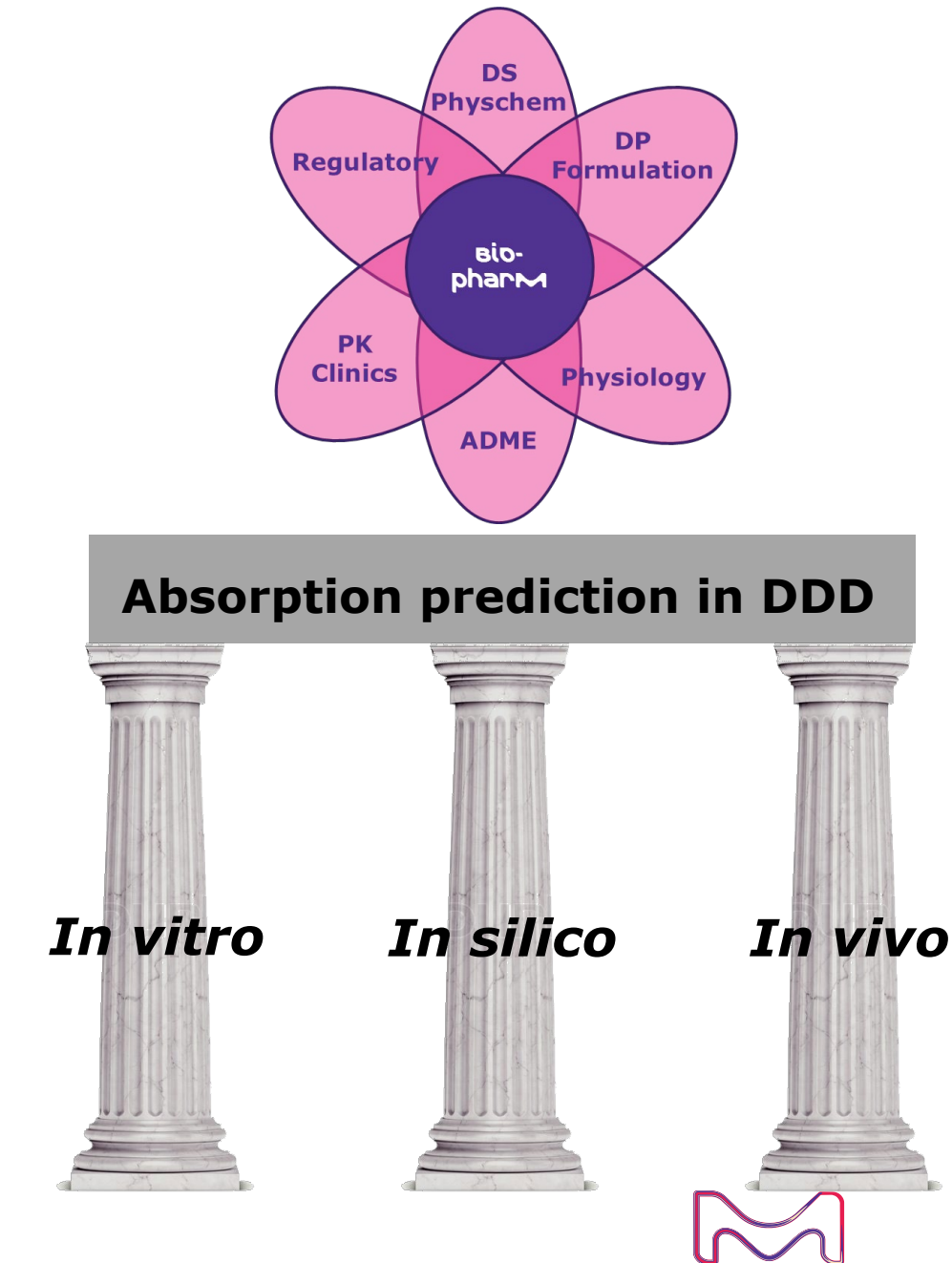
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## Conclusion

Building absorption models with high confidence:

We are on a good way, but we are not yet there – especially for drugs with low to moderate absorption

- What is the most predictive biorelevant medium?
- Scaling of precipitation from *in vitro* to *in vivo*
- Permeability aspects, esp. active transport
- Shortcomings in current PBPK software



# Thank you for your attention 😊

## Dr. Christian Wagner

Head of CoE Biopharmaceutics

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